# LETTER

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# Letter

# Improved photoemission and stability of 2D organic-inorganic lead iodide perovskite films by polymer passivation

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### Abstract

2D organic-inorganic lead iodide perovskites hold great promise for functional optoelectronic devices. However, their performances have been seriously limited by poor long-term stability in ambient environment. Here, we perform a systematic study for the stability improvement of a typical 2D organic-inorganic lead iodide perovskite (PEA)<sub>2</sub>PbI<sub>4</sub>. The degradation of the (PEA)<sub>2</sub>PbI<sub>4</sub> films can be attributed to the interaction with the humidity in environment, which leads to decomposition of the perovskite components. Then, we demonstrate that polymer passivation provides an effective approach for improving the crystal quality and stability of the (PEA)<sub>2</sub>PbI<sub>4</sub> films. Correspondingly, the photoemission of the polymer-passivated (PEA)<sub>2</sub>PbI<sub>4</sub> films has been enhanced due to the decreased trap states. More importantly, a hydrophobic polymer (Poly(4-Vinylpyridine), PVP) will protect the (PEA)<sub>2</sub>PbI<sub>4</sub> films from humidity in ambient environment, which can greatly improve the physical and chemical stability of the 2D perovskite films. As a result, the PVP-passivated (PEA)<sub>2</sub>PbI<sub>4</sub> films can produce a bright emission even after long-term (>15 d) exposure to ambient environment (25 °C, 80% RH) and continuous UV illumination. This work provides a convenient and effective approach for improving the long-term stability of 2D organic-inorganic lead iodide perovskites, which shows great promise for fabricating large-area and versatile optoelectronic devices.

Supplementary material for this article is available online

Keywords: 2D perovskites, photoemission, stability, polymer passivation, optoelectronic devices

(Some figures may appear in colour only in the online journal)

## 1. Introduction

Recently, organic-inorganic hybrid perovskites have shown great promise for functional optoelectronic devices including solar cells, light-emitting devices (LEDs), and optoelectronic

<sup>&</sup>lt;sup>4</sup> Contributed equally to this work

detectors [1–7]. Organic-inorganic perovskites have a high band-edge absorption, a low defect density and a long carrier diffusion length [8–10]. The extraordinary optoelectronic characteristics and solution processability make it suitable for fabricating large-area and high-performance devices [11–13]. In particular, 2D organic-inorganic hybrid perovskites with a layered multi-quantum-well structure, have been demonstrated to possess a stronger quantum confinement [14–16] and a greater tenability [17–19]. Due to the more promising advantages compared to their 3D counterparts, 2D organicinorganic hybrid perovskites have become a great candidate for next-generation optoelectronic devices [20–2627].

Although they have shown outstanding optical properties, organic-inorganic hybrid perovskites still suffer from poor stability caused by moisture, oxygen, interface defects and light illumination [28–33]. The instability will lead to a degradation of the perovskite in ambient environment, which has seriously limited the lifetime and performance of the optoelectronic devices based on perovskites. To overcome the shortage caused by the instability, great efforts have been made using surface passivation. For example, the stability of the 3D hybrid perovskite films could be enhanced by modifying the surface with crosslink groups [34, 35] and the surface defects can be passivated by using organic halide salt [36, 37]. More recently, potassium halide layers were demonstrated to be effective for passivating the perovskite interfaces and achieving highstable photoluminescence (PL) emission [38]. However, for 2D organic-inorganic hybrid perovskites, the concerns about the stability still remains to be investigated in detail, which has great importance for fabricating high-performance functional devices.

In this work, we perform a systematic study on the degradation and stability improvement of a typical 2D organicinorganic lead iodide perovskite (PEA)<sub>2</sub>PbI<sub>4</sub>. The degradation of the (PEA)<sub>2</sub>PbI<sub>4</sub> films can be attributed to a decomposition of the perovskite components, especially by interacting with the humidity and oxygen in the environment. Through surface passivation with a hydrophobic polymer, the surface traps of the (PEA)<sub>2</sub>PbI<sub>4</sub> films can be decreased. More importantly, the hydrophobic polymer will protect the hybrid perovskite from humidity in the environment and greatly improve the stability of the (PEA)<sub>2</sub>PbI<sub>4</sub> films. As a result, the polymer-passivated perovskite films can produce a bright emission even after longterm (>15 d) exposure to ambient environment and continuous UV illumination, which shows great promise for functional optoelectronic devices such as optoelectronic displays.

#### 2. Result and discussion

(PEA)<sub>2</sub>PbI<sub>4</sub> films were prepared using a one-step spin-coating method (supporting information) [14]. After that, the films were systematically characterized by optical and morphologic measurement. Figures 1(a) and (b) show the PL image and SEM image of the prepared (PEA)<sub>2</sub>PbI<sub>4</sub> film respectively. The PL image displays a bright and homogenous green emission, implying the good quality of the (PEA)<sub>2</sub>PbI<sub>4</sub> film. The SEM image shows that the 2D perovskite film consists of well-defined grains characterized as large crystals with average sizes between 5 ~ 15  $\mu$ m. The XRD pattern in figure 1(c) displays well-defined diffraction patterns corresponding to the (00 l) series planes, and the narrow and sharp peaks imply a high quality of the perovskite film. Furthermore, figure 1(d) shows the absorption spectrum of an as-prepared (PEA)<sub>2</sub>PbI<sub>4</sub> film, which agrees well with the reported results [39, 40]. The absorption spectrum displays a narrow absorption peak located at ~ 517 nm, which corresponds to the intrinsic exciton absorption of (PEA)<sub>2</sub>PbI<sub>4</sub> [18].

To investigate the degradation of the  $(PEA)_2PbI_4$  films, the PL emission was monitored over time, as shown in figure 2(a). The samples were placed in ambient environment (25 °C, 80% RH), under continuous UV-light illumination. At first, the asprepared  $(PEA)_2PbI_4$  film shows a bright green light, which is due to the exciton recombination of the 2D perovskite excited by high-energy photons. However, the green light becomes weaker over time and is completely quenched after about 4 d. To quantitatively characterize the degradation process, the PL intensity was extracted and plotted as a function of time, as shown in figure 2(b). We can observe that the PL intensity will decrease to 50% of the initial intensity after ~ 2 d, which shows direct evidence of the serious degradation of the 2D perovskite film.

When exposed to ambient atmospheric conditions such as humidity, air and light illumination, the organic-inorganic hybrid perovskite will be degraded due to a series of reactions. To reveal the degradation mechanism of the (PEA)<sub>2</sub>PbI<sub>4</sub> films, the absorption spectrum of the film was measured over time. Figure 2(c) shows that the intensity of the exciton absorption peak largely decreases as time increases, and the absorption peak nearly disappears after 9 d (figure 2(d)), which is consistent with the PL measurement result. It is worth noting that, a new absorption peak located at ~ 410 nm appears in the absorption spectra, with the intensity growing gradually over time. The appearance of this absorption peak provides a strong evidence for the presence of PbI4<sup>2-</sup> in the film during degradation [41]. The generation of  $PbI_2$  in the organic-inorganic lead iodide perovskite films can be attributed to the following process [30]

$$(PEA)_2PbI_4 \rightarrow PbI_2 + 2 \left[ PEA^+ + I^- \right] \rightarrow (PbI_2)_{solid} + 2PEA_{gas} + 2HI_{gas}$$
(1)

Moreover, tri- and tetra-iodoplumbate complexes will be produced due to the reaction between  $PbI_2$  and I<sup>-</sup>, which can be expressed as

$$PbI_2 + I^- \rightarrow PbI_3$$
 (2)

$$PbI_3 + I^- \to PbI_4^2 \tag{3}$$

To further confirm the degradation process, XRD of the  $(PEA)_2PbI_4$  film was measured at various times, as shown in figure 2(e). After 5 d, the XRD peaks corresponding to the pristine  $(PEA)_2PbI_4$  shift slightly to larger diffraction angles, indicating that the perovskite grains have started to degrade



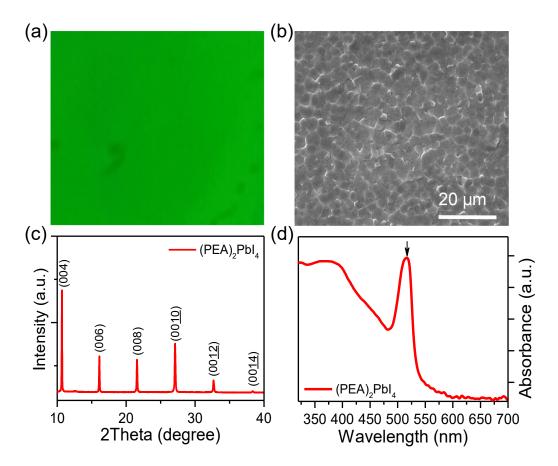
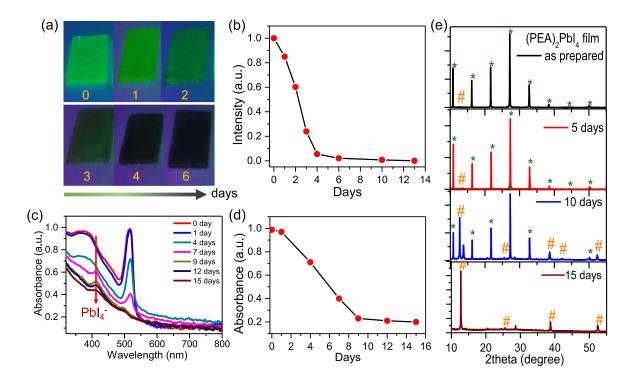
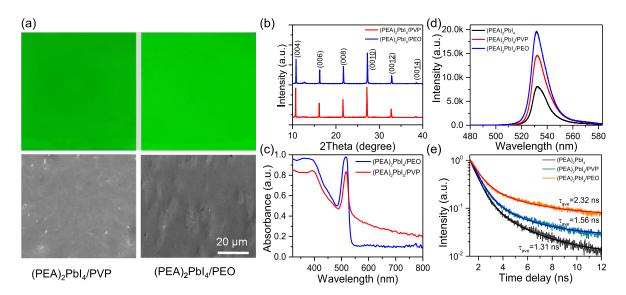


Figure 1. (a) PL image, (b) Top-view SEM image, (c) XRD pattern and (d) absorption spectrum of a pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film.



**Figure 2.** (a) Monitored PL images of a pristine  $(PEA)_2PbI_4$  film over time. (b) Plot of corresponding PL intensities extracted from the PL images. (c) Monitored absorption spectrum of a pristine  $(PEA)_2PbI_4$  film over time. (d) Plot of the exciton absorption peak intensity as a function of time. (e) XRD pattern of a pristine  $(PEA)_2PbI_4$  film measured over time.





**Figure 3.** (a) PL images (top panel) and SEM images (bottom panel) of (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films, respectively. (b) XRD patterns and (c) absorption spectra of the as-prepared (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films respectively. (d) Measured PL spectra and (e) corresponding TRPL traces for the (PEA)<sub>2</sub>PbI<sub>4</sub>, (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films respectively.

[30, 42]. Moreover, after 10-day exposure under UV light, a peak located at  $12.72^{\circ}$  increased significantly, with appearance of more new peaks located at  $25.55^{\circ}$ ,  $38.7^{\circ}$ ,  $44.3^{\circ}$  and  $52.34^{\circ}$  (labelled by #). Previous reports demonstrate that the new peaks can be attributed to the production of PbI<sub>2</sub> in the perovskite film [43–45], which further confirm the degradation process expressed by equation (1). In particular, after 15 d, all of the XRD peaks corresponding to the pristine (PEA)<sub>2</sub>PbI<sub>4</sub> disappear, suggesting complete degradation of the 2D perovskite film.

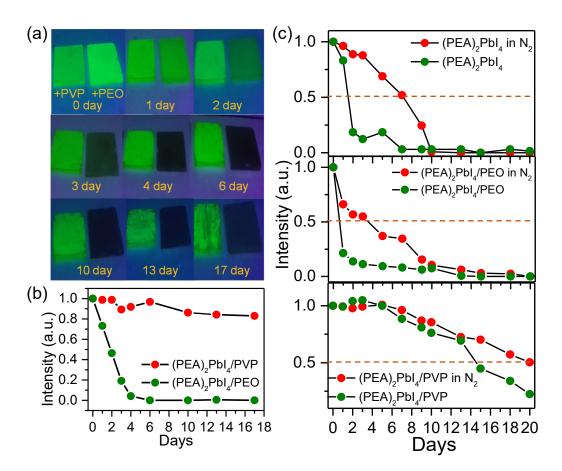
Surface passivation and encapsulation have shown great promise for effectively suppressing the degradation of the organic-inorganic perovskite in ambient environment [46–48]. To improve the stability of the (PEA)<sub>2</sub>PbI<sub>4</sub> film, Poly(4-Vinylpyridine) (PVP) and Poly(ethylene oxide) (PEO) were used for encapsulation and surface passivation. The polymer solution was respectively coated on the perovskite films, forming the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO composites. Figure 3(a) presents the PL images (top panel) and SEM images (bottom panel) of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films respectively. The PL emission of the polymer encapsulated films has been increased compared with the intrinsic (PEA)<sub>2</sub>PbI<sub>4</sub> film. After polymer deposition, the surface morphology of the films becomes smooth and exhibit no pinholes on the surface, especially by utilizing PVP layer, indicating a modification of the film quality. Moreover, polymer encapsulation can heal the defect states on the surface of the hybrid perovskite films, thus leading to a suppressed nonradiative recombination and an enhanced PL emission compared with the pristine perovskite film [12, 36, 49]. Figures 3(b) and (c) display the XRD patterns and absorption spectra of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films respectively. The consistence of the XRD patterns and absorption spectra of the pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film and the polymer-coated ones suggests that the polymer passivation will not affect the crystal structure and bandgap of the (PEA)<sub>2</sub>PbI<sub>4</sub> film.

**Table 1.** Fitted parameters for the PL decay traces of pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film, (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film. The average lifetime is calculated as  $\tau_{ave} = (A_1\tau_1^2 + A_2\tau_2^2)/(A_1\tau_1 + A_2\tau_2)$ .

	(PEA) <sub>2</sub> PbI <sub>4</sub>	(PEA) <sub>2</sub> PbI <sub>4</sub> /PVP	(PEA) <sub>2</sub> PbI <sub>4</sub> /PEO
$\overline{A_1}$	4019.5	3650.2	2688.3
$\tau_1$ /ns	0.674	0.735	0.788
$A_2$	429.5	483.6	678.1
$\tau_2$ /ns	2.767	3.12	3.63
$\tau_{ave}$ /ns	1.31	1.56	2.32

To quantitatively characterize the photoemission performances of the (PEA)<sub>2</sub>PbI<sub>4</sub> films encapsulated by PVP and PEO, PL spectra and lifetime were measured using a µ-PL system. The samples were pumped with a femtosecondpulsed Ti:sapphire laser (800 nm, ~ 8 fs, 80 MHz), the infrared photons from which are beneficial for protecting the perovskite films from damage [18, 50-53]. Figure 3(d) displays the measured PL spectra of the (PEA)<sub>2</sub>PbI<sub>4</sub> films, under a pumping power of 2 mW (~2.6 kW/cm<sup>2</sup>). As the photon energy of the femtosecond laser is smaller than the bandgap of (PEA)<sub>2</sub>PbI<sub>4</sub>, the emission can be attributed to a two-photon absorption induced PL (figure S1, supporting information, available online at stacks.iop.org/NANO/31/42LT01/mmedia). Notably, the PL intensity of the polymer-passivated (PEA)<sub>2</sub>PbI<sub>4</sub> films is stronger than that of the pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film, which agrees well with the optical images in figure 3(a). Figure 3(e)presents the PL decay traces for the pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film, (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film, and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film respectively. As reported, the PL emission of the 2D hybrid perovskites can originate from both the intrinsic exciton recombination and the trap states recombination. To reveal the two processes, all the decay traces can be fitted to a biexponential decay function  $I(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2)$ . The fitted parameters





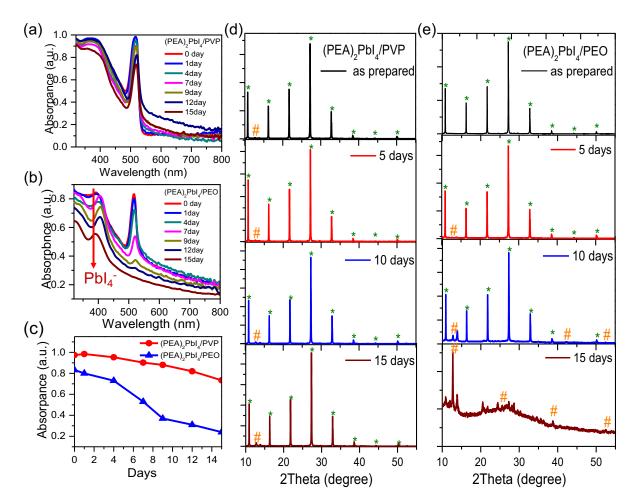
**Figure 4.** (a) Monitored PL images of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films over time. (b) Plot of the extracted PL intensity of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films as a function of time. (c) Plot of the extracted PL intensity of the (PEA)<sub>2</sub>PbI<sub>4</sub>, (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films in ambient and in N<sub>2</sub>, as a function of time.

are presented in table 1. In particular, the average lifetimes of the 2D perovskite films increase from 1.31 ns to 1.56 ns and 2.32 ns for (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film respectively. The increased PL lifetimes indicate that the nonradiative recombination is decreased in the polymerencapsulated perovskite films, which is consistent with the PL enhancement and further demonstrates a suppression of the surface trap states [54, 55].

To investigate the effect of different polymer passivation on the stability of the (PEA)<sub>2</sub>PbI<sub>4</sub> films, the samples were kept in ambient environment and exposed to continuous UV-light illumination. Figure 4(a) presents the PL images of the polymerencapsulated films over time. One can observe that the PL emission of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film quickly decreases, while the emission of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film remains stable as time increases. To quantitatively characterize the PL variation, the intensity was extracted and plotted as a function of time, as shown in figure 4(b). For the (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film, the PL intensity has been decreased to less than half of the initial intensity after 3 d, and it becomes completely quenched after 4 d, indicating that the samples have been damaged under UV light-irradiation. In contrast, the PL intensity of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film has only decreased by 20% even after 17-days. Though the (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film is isolated from air in the ambient environment, the degradation property is the same as the intrinsic (PEA)<sub>2</sub>PbI<sub>4</sub> film, demonstrating that humidity is the dominant factor which leads to the degradation of the organic-inorganic hybrid perovskites, especially exposed to continuous light illumination. In addition, we also monitored the degradation of the (PEA)<sub>2</sub>PbI<sub>4</sub> films placed in a bag filled with dry  $N_2$  (figure 4(c) and figure S2, supporting information). The degradation of the pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film has been greatly slowed down, demonstrating that the degradation can be dominantly attributed to the humidity in the ambient environment. As PEO is a hydrophilic polymer, it can absorb the water in the environment, further aggravating the degradation of the (PEA)<sub>2</sub>PbI<sub>4</sub> film in humidity. In contrast, PVP is a hydrophobic polymer, which will prevent the (PEA)<sub>2</sub>PbI<sub>4</sub> film from damage by the air and humidity from the environment. As a result, the PL emission from the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film remains nearly unchanged over time, and it is still quite bright even after a long time of 17 d. This result implies a great improvement of the stability of the 2D perovskite film for light illumination and air/humidity exposure by PVP polymer passivation.

To further demonstrate the stability improvement by PVP passivation, the absorption spectra of the pristine and polymerencapsulated (PEA)<sub>2</sub>PbI<sub>4</sub> films were measured over time, as shown in figures 5(a) and (b). The intensity at the exciton





**Figure 5.** Absorption spectra of the (a) (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (b) (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film, measured as a function of time. (c) Plot of the exciton absorption peak intensity of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO films. Measured XRD patterns of (d) (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (e) (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film, as a function of time.

absorption peak was extracted and plotted in figure 5(c). Similar to the fluorescence result, the absorption spectrum of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film remains stable and the exciton absorption peak decreases to 80% of the initial intensity after 15 d. However, the absorption spectrum of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film exhibit a large decrease over a wide range, especially for the exciton absorption peak located at ~ 516.5 nm, which provides evidence for the degradation of the perovskite grains. In particular, the exciton absorption peak of the perovskite films disappear after 9 d, indicating that the perovskite has been completely damaged in the ambient environment. The stability of the (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film can be also revealed by the XRD measurement results, as shown in figures 5(d) and (e), respectively. Furthermore, a decreased PVP concentration exhibits a similar effect on the perovskite passivation, indicating an excellent performance of the PVP passivation layer (figure S3, supporting information).

Organic-inorganic lead iodide perovskite films are convenient for fabricating into large-area structures and patterns, which make them excellent candidates for practical applications in optoelectronic displays. To demonstrate the great promise of the polymer-passivated 2D perovskite films for functional applications, the  $(PEA)_2PbI_4$  solution was written into a pattern spelling 'HUST', and then coated with polymer films. The PL emission from the films were monitored under continuous UV illumination, as shown in figure 6. One can observe that the pattern passivated by PVP can produce a clear and bright emission for a much longer time (>15 d) than the pristine and PEO-passivated (PEA)<sub>2</sub>PbI<sub>4</sub> films, indicating their great potential for light displays. Moreover, the (PEA)<sub>2</sub>PbI<sub>4</sub> film can be well protected by the PVP passivation layer, which is effective for increasing the physical and chemical stability of the perovskite films. Although some other materials such as 2D materials have also been proposed to suppress the degradation of hybrid perovskites due to their excellent stability [30, 56, 57], it is still difficult to efficiently cover the atomic-thin layers of the perovskite films. Therefore, the PVP-passivated 2D hybrid perovskite films shows great promise for large-area and versatile optoelectronic devices.

#### 3. Conclusion

The stability issue of  $(PEA)_2PbI_4$  films was investigated by PL emission, absorption spectrum and XRD in ambient environment. The degradation mechanism of the  $(PEA)_2PbI_4$  films



**Figure 6.** Monitored PL images of 'HUST' patterns, composed of pristine (PEA)<sub>2</sub>PbI<sub>4</sub> film, (PEA)<sub>2</sub>PbI<sub>4</sub>/PVP film and (PEA)<sub>2</sub>PbI<sub>4</sub>/PEO film respectively.

was revealed to be the interaction with humidity, which leads to decomposition of the perovskite components. More importantly, a hydrophobic polymer (PVP) encapsulation provides an effective approach for passivating surface traps and isolating the water from the environment. As a result, the PVPpassivated 2D organic-inorganic hybrid perovskite films can produce a bright emission even after long-term (>15 d) exposure to ambient environment and continuous UV illumination. This work provides a convenient and effective approach for improving the long-term stability of 2D organic-inorganic lead iodide perovskites, which shows great promise for fabricating large-area and versatile optoelectronic devices.

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